

## Qualitative Analysis (MCQ)

1. Four pairs of solutions are mixed.

Which pair of solutions forms a white precipitate?

- A  $\text{NH}_4\text{Cl}(\text{aq})$  and  $\text{NaOH}(\text{aq})$
- B  $\text{KBr}(\text{aq})$  and  $\text{AgNO}_3(\text{aq})$
- C  $\text{FeCl}_3(\text{aq})$  and  $\text{NH}_3(\text{aq})$
- D  $\text{Cr}_2(\text{SO}_4)_3(\text{aq})$  and  $\text{BaCl}_2(\text{aq})$

Your answer

[1]

END OF QUESTION PAPER

## Mark scheme – Qualitative Analysis (MCQ)

Question			Answer/Indicative content	Marks	Guidance
1			D	1	
			<b>Total</b>	<b>1</b>	